

# Snow atmospheric fluxes of reactive nitrogen species in the European high Arctic

Antonietta Ianniello<sup>1</sup>, Francesca Spataro<sup>1</sup>, Rosamaria Salvatori<sup>1</sup>, Mauro Valt<sup>2</sup>, Giulio Esposito<sup>1</sup>, Mauro Montagnoli<sup>1</sup>

<sup>1</sup>CNR - Institute of Atmospheric Pollution Research, Via Salaria Km 29.3, 00015 Monterotondo S., Rome, Italy. E-mail address: ianniello@iia.cnr.it

<sup>2</sup>ARPAV – Arabba Avalanche Centre, Via Pradat 5, 32020 Arabba di Livinallongo, Belluno, Italy

#### Introduction

Atmospheric nitrogen oxides (NO<sub>x</sub> = NO + NO<sub>2</sub>), emitted from fossil fuel combustion, biomass burning, soil microbial activity, and lightning with a small contribution from downward transport from the stratosphere and high-flying aircraft, have adverse respiratory effects, contribute to the formation of atmospheric acidity, and play a pivotal role in regulating the oxidative capacity through their connections to ozone (O<sub>3</sub>) and hydroxyl radical (OH).

Previous studies in the Arctic, Antarctic, and at mid-latitudes have demonstrated that the sunlit polar snowpack is an active reactor, where previously deposited chemical species can undergo photochemical processes that subsequently result in an efflux of trace gases to the overlying atmosphere, altering its composition and oxidative capacity with important implications for climate processes, including atmospheric oxidation and aerosol formation. In particular, the photochemical production of nitrogen oxides (NO<sub>x</sub>) and nitrous acid (HONO) by nitrate (NO<sub>3</sub><sup>-</sup>) photolysis within or at the surface of snowpack can significantly impact the NO<sub>x</sub>, radical species (OH<sub>x</sub>), and ozone  $(O_3)$  budgets in the overlying atmosphere and alter the preservation and concentration of trace species such as hydrogen peroxide ( $H_2O_2$ ) and  $NO_3^-$  in ice cores (Beine et al. 2006; Björkman et al. 2014; Ianniello et al. 2016; Spataro et al. 2016). Factors known to affect the release of chemicals include the physical properties of snow, such as temperature, pH, ionic strength and specific surface area (SSA). However, the quantitative understandings of NO<sub>x</sub> emissions from snow is still incomplete, but it is a research priority to be able to parameterize global chemistry climate models to assess for example global impacts of chemical air-snow exchange on tropospheric O<sub>3</sub>. In addition, knowledge of the natural background concentration and main sources of reactive nitrogen species ( $NO_y = NO_x + HNO_3 + HONO + particulate NO_3 + PAN + etc.$ ) is pivotal in determining their influence and the impact of human activity on the atmospheric photochemistry and on ecosystems in polar environments. Observations on atmospheric chemistry of  $NO_v$  in the global atmosphere are limited, in particular, in the polar regions due to limited sensibility of sampling techniques, and to a general lack of comparable measurements of  $NO_x$  emissions from surface snow across a wide range of environmental conditions. Hence, a better understanding of these processes controlling the fate of  $NO_v$  in the polar regions is needed.

As part of the PRIN2007 ("Dirigibile Italia"), PRIN2009 ("Arctica") and ARCA projects, measurements of atmospheric concentrations and fluxes of reactive nitrogen species (NO, NO<sub>2</sub>, NO<sub>x</sub>, HONO, HNO<sub>3</sub> and particulate NO<sub>3</sub>"), and chemical and physical properties of snow were carried out during spring time 2010 and 2011 at Ny-Ålesund (Svalbard Islands). Furthermore, automatic and continuous measurements of nitrogen oxides were performed throughout the year 2015.

The goals of this work were to determine the atmospheric concentrations, the partitioning and the budget of  $NO_y$ ; to quantify the direction and magnitude of  $NO_y$  fluxes at the snow-atmosphere interface; to identify chemical processes that control the  $NO_y$  redistribution and recycling in the snow; and, then, to investigate about the atmospheric sources of snow  $NO_3^-$  and the links with climate changes.



## Experimental

All measurements were carried out from 29 March to 30 April 2010 and from 1 to April 2011 in a flat and undisturbed snow area near the Gruvebadet station, located at about 1 km south from the Ny-Ålesund International Arctic Research and Monitoring Facility, Svalbard (78°55'N, 11°54'E, 40 m msl). All the instruments had been located in a laboratory hut. This measuring site was chosen to minimize the influence of local pollution from the research community. The snowpack flux experiments were derived from independent chemical measurements of reactive nitrogen species at two sampling heights above the snow (0.3 and 1.5 m) using a multi-level gradient grid (scaffolding) tower at height of 10 m above the snow surface. This grid tower was installed at the site before the onset of the snow accumulation season and allowed to be covered by the precipitating snow without any disturbance throughout the snow-covered time period. Simultaneous atmospheric turbulence measurements were made using a 3-dimensional sonic anemometer in order to determine the friction velocity (u\*) and eddy diffusivity between the reactive nitrogen fluxes from this mixed eddy covariance and gradient technique is fully described previously (Beine et al. 2003, Ianniello et al. 2016).

 $NO_x$  were detected every 6 min using a modified commercial two-channel high-sensitivity chemiluminescence detector (Sonoma Technologies, USA). Channel 1 sampled air at 0.3 m and channel 2 at 1.5 m above the snow in order to determine the concentration gradients of the nitrogen species. The 3 $\sigma$  detection limit for NO was 4.4 and 3.3 ppt for Channel 1 and 2, respectively, while the 3 $\sigma$  detection limit for NO<sub>2</sub> was 9 and 7 ppt for Channel 1 and 2, respectively, in a 1 min average.

Nitrous acid (HONO) was measured every 9 min by two-channel system with derivatization and analysis by HPLC equipped with UV-Vis absorption detector. The  $3\sigma$  detection limit for HONO was 0.5 ppt (Spataro et al. 2016).

Measurements of HNO<sub>3</sub>, and fine (< 2.5  $\mu$ m) and coarse (2.5 >  $\mu$ m) particulate NO<sub>3</sub><sup>-</sup> were carried out with 12 h and 24 h sampling times at two heights above the snow by two independent annular denuder lines (Ianniello et al. 2007; 2016).

In addition, determination of  $O_3$  and chemical and physical properties of snow (ionic composition, pH, temperature, specific surface area) were carried out. The specific surface area (SSA) was determined by short-wave infrared (SWIR) method (Dominé et al. 2006; Salzano et al. 2016). All snow and denuder samples were analysed on site by ion chromatography for major anions and cations within 24 h of sampling.

Further, measurements of  $NO_x$  and  $O_3$  were performed every 1 min using automatic and commercial analysers from 18 May to 8 October 2015 at the Gruvebadet station.

### Results

Both NO and NO<sub>2</sub> showed clear diurnal cycles with noontime maxima and nighttime minima during spring time 2010 and 2011 (Figure 1). Significant emission fluxes of NO and NO<sub>2</sub> were observed, reaching noontime values up to 19.42 and 25.20 pmol/m<sup>2</sup> s, respectively (Figure 2). The snow surface was the source of NO and NO<sub>2</sub> but these observed releases were small due to almost alkaline snow environment and chemical forms of snow NO<sub>3</sub><sup>-</sup>. Nitrate concentrations in the snowpack were variable with the highest values during the snowfall events and were related to the specific surface area of the snow indicating that NO<sub>3</sub><sup>-</sup> was located on surface snow. In fresh and acid snows, NO<sub>3</sub><sup>-</sup> was present as dissolved or adsorbed HNO<sub>3</sub> taken up from the gas phase. This suggested that snow NO<sub>3</sub><sup>-</sup> was available for photochemical reactions and for physical exchanges, that led to NO and NO<sub>2</sub> emissions at Ny-Ålesund. However, the presence of sea salt and crustal particles in this marine environment made also the snow alkaline. Indeed, the ionic composition of snow revealed that NO<sub>3</sub><sup>-</sup> in aged snows was associated with heavier cations such as Na<sup>+</sup>, and, then, was not always reactive regard to physical exchanges with the atmosphere, resulting in low NO and NO<sub>2</sub> emissions.



No clear diurnal cycle for the HONO concentrations was observed. Emission of HONO from the snow surface was only a minor HONO source in the atmosphere at Ny-Ålesund. Significant deposition fluxes of HNO<sub>3</sub>, fine and coarse particulate NO<sub>3</sub><sup>-</sup> were also observed, reaching peak values up to -18.00, -37.80 and -12.50 pmol/m<sup>2</sup> s, respectively, during snowfall events and contributed up to 24 % of the NO<sub>3</sub><sup>-</sup> (–N) budget in the snow. However, these dry depositions were not large at Ny-Ålesund, so that the main NO<sub>3</sub><sup>-</sup> source for snow was wet deposition in falling snow. In addition, emissions of HNO<sub>3</sub> were also observed during windy periods and were related to a decrease in snow nitrate concentration.

Furthermore, during the time period 2015 the measurement results showed that long-range transport and local air pollution impact on the NO and NO<sub>2</sub> concentrations as observed during spring season 2010 and 2011. Air coming from NW and NE directions had higher values of NO<sub>x</sub> during spring and summer periods. The highest mean concentration observed when wind was from northerly direction is consistent with the influence of local pollution sources on NO<sub>x</sub> measurements.

#### References

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concentrations and fluxes together with UV radiation during spring season 2010

